in order to make an empirical assessment of the *relative* powers of the traditional and reinforced inequalities. It was then found that the reinforced criterion can give definite signs and several exclusions of sign combinations, even when the traditional inequalities gave no information.

If the known atoms are in positions having the same symmetry as the rest of the structure, no particular type of inequality is expected to be especially affected. In the case where the atoms with known positions define a higher symmetry than the rest, certain inequalities will be reinforced to a greater extent than others. However, since it would be natural to calculate

$$\sum_{n}' g f_{(n)} \cdot \exp\left[2\pi i \mathbf{h} \cdot \mathbf{r}_{(n)}\right]$$

for all h's simultaneously, there is little point in discussing particular situations in detail. It is evident that a sum,

Acta Cryst. (1961). 14, 1204

$\sum_{n}' g f_{(n)} \cdot \exp\left[2\pi i \mathbf{h} \cdot \mathbf{r}_{(n)}\right]$,

that is large, whatever the cause might be, is particularly capable of increasing the power of an inequality.

It seems specially valuable to consider the reinforcement proposed here in the case in which some atomic positions are determined by symmetry, because

$$\sum_{n} g f_{(n)} \cdot \exp\left[2\pi i \mathbf{h} \cdot \mathbf{r}_{(n)}\right]$$

is then exactly known, and V very easily found from U.

References

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Löfgren, T. (1961). Acta Cryst. 14, 434.

Cell parameters and space groups of potassium, rubidium, and cesium acid chloromaleates.*

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The cell parameters and space groups of the potassium, rubidium, and cesium acid chloromaleates

(HOOC.CCl:CH.COOM, where M = K, Rb, or Cs)

grown from aqueous solution were determined. Preliminary values of the cell constants as well as symmetry information were found from photographs prepared with Mo $K\alpha$ radiation on the precession camera. Powder diffraction data taken with a Debye–Scherrer camera and Cr $K\alpha$ radiation were used in subsequent leastsquares refinements of the cell dimensions. Densities were measured pycnometrically; the supporting liquid was *n*-decane.

Potassium acid chloromaleate was found to be orthorhombic. Precession camera pictures indexed as h0l, h1l, h2l, 0kl, 1kl, 2kl, and hk0 display the following systematic absences: none among reflections hkl; hk0 reflections present only if h+k=2n; h0l, only if l=2n; and 0kl, only if k=2n. These absences indicate the space group

* Work performed for the U.S. Atomic Energy Commission at the Oak Ridge National Laboratory, operated by the Union Carbide Corporation, Oak Ridge, Tennessee. Pbcn. A previous report (Yardley, 1925) gave cell dimensions in good agreement with those of this study, but a different space group (*Pmcn*, referred to the present axis orientation). The disagreement rests on two weak reflections which are absent on our photographs.

The space group of the two isomorphous monoclinic salts rubidium and cesium acid chloromaleate was not uniquely determined. On the precession camera pictures hk0, hk1, hk2, 0kl, 1kl, and 2kl, the only systematic absences noted were those for which h + k + l = 2n + 1. Of the three space groups Im, I2 and I2/m, which are consistent with these absences, the last is considered unlikely because it requires that two chloromaleate ions lie in mirror planes in the approximately five by eight Ångström face of the cell.

The findings of this study are summarized in Table 1. In the monoclinic cases, the more usual *C*-centered cell is included, but the body-centered cell with β close to 90° is retained for convenience. The observed interplanar spacings, corrected for error due to absorption in the specimen (Klug & Alexander, 1954), are compared with the calculated values in Tables 2, 3, and 4. Indexing of

Table 1. Crystallographic data for potassium, rubidium, and cesium acid chloromaleates

		Rubidi	um	Cesium		
	Potassium	Body-centered indexing	C-centered indexing	Body-centered indexing	C-centered indexing	
Crystal system	Orthorhombic	Monoclinic	Monoclinic	Monoclinic	Monoclinic	
a b c	$\begin{array}{c} 15{\cdot}815\pm0{\cdot}015 \ \text{\AA} \\ 10{\cdot}928\pm0{\cdot}006 \\ 7{\cdot}707\pm0{\cdot}005 \end{array}$	$\begin{array}{c} 8{\cdot}112\pm0{\cdot}003 \ \text{\AA} \\ 16{\cdot}638\pm0{\cdot}011 \\ 5{\cdot}090\pm0{\cdot}003 \end{array}$	9·61 Å 16·64 5·09	$\begin{array}{c} 8{\cdot}352\pm0{\cdot}007 \ \text{\AA} \\ 17{\cdot}325\pm0{\cdot}014 \\ 5{\cdot}152\pm0{\cdot}004 \end{array}$	9·87 Å 17·33 5·15	
etaSpace group Molecules/cell	Pbcn 8	$\begin{array}{c} 90^\circ\ 28'\pm4'\\ Im \ {\rm or}\ I2 \end{array}$	122° 26' Cm or C2 4	$\begin{array}{c} 90^{\circ} \ 45^{\prime} \pm 5^{\prime} \\ Im \ \text{or} \ I2 \end{array}$	122° 12' Cm or C2 4	
Calculated density Observed density	1·881 g.cm. ⁻³ 1·868	$\begin{array}{c} 2 \cdot 272 \\ 2 \cdot 246 \end{array}$	g.cm. ⁻⁸	2.516 g.cm, ⁻³ 2.515		

The errors listed are least-squares standard deviations.

SHORT COMMUNICATIONS

Spacings for K, Rb, Cs acid chloromaleates

Table 2. K H Cl Maleate

1

VW- •

М

•

4

W+

М

М+

Ψ+ .

v₩-- Ъ

ΥΨ - b

γ₩-ь

M+

h

2

0

6

6

6

k 1

2

1

30

3

20 2.229

2 2 2,207

2

2 2.145

2 2.021

2

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0

2

6

1

a

2

55

1

3 2 6 0 1.775

The observed spacings have been corrected for absorption error. Specimen radius = 0.075 mm.

0 1.897

٥

ŀ 4

n

d, Å

2.230

2.212

2.200

2,176

2.165

2.134

2.127

2.127

2.114

2.084

2 081

2.058

2.053

2.032

2.030

2.029

2 019

2.015

2.004

1.997 1.977

1.962

1.953

1.950

1.949 1.941

1.927

1.920

1.919

1.913

1.888 1.886

1.884 1.873

1.872

1.872 1.868

1.863 1.859 1.859

1.854 1.848 1.845

1.842

1.836

1.822

1.821

1.821 1.817

1.814

1.810

1.807

1.80 0 1.798

1.788

1.785

1.773

1.859 w_

R, 1.788

d., Å

9.013

7.923

4.915

4.742

4.299 s

3.884

3,744 S-

3.540 M4

3.469

3,297

3.231 W. ь•

3.149 s

3.091 М

3.036

2.994 М

2,958 M+

2 792

2.760

2.704 ₩-

2.629

2.525

2.472

2.371 w--

2.326

2.279 w ь

s.990 7.907

5.851

4.927 4.748 4.495 4.457 4.290

4.042

3.954

3.883

3.854

3.744

3.550

3.542

3.518

3.464

3.404

3.349

3.302

3.224

3.203

3.149

3.111

3.088

3.040

2.997

2.058

2.926

2.827 2.793

2.737 2.732

2.703

2.676

2.630 2.611 2.582 2.580

2.510

2.463 2.448

2.325

2.300

2.259

2.232

Indicates a broad line,

V = very, S = strong, M = medium, W = weak.

0 0

0 5.464 4.927

n

1. 1 , d ., Å

1 1

1 1 1

02 2

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0 2 2.445 2.431

2 0

3

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> 1 2.575

1 2 2,530

3 2

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2 2.386 2.384

2 · 2.360

3

1 2.269

ō 2,248

d_c, Å h k 1 d_, Å Ь k 1 d_, Å d_o, Å 1_ 1, 0 2 0 8.319 8.293 7.281 W W-٥ 2,380 2.230 W . 2.335 1 7.291 4.867 4.861 6 1 -1 2.330 tr ₩+ 4.578 4.582 . 2 2.302 3 0 2.301 2.291 -1 -1 5 2 4.312 ₩+ õ 4.296 1 2.170 Мн ō 2.289 0 4.159 0 ñ 4.056 4.056 tr •ь 3 27 1 2.281 3.839 3.817 3.750 0 2.146 w... 2 3.829 M+ ь -2 2.229 2 1 3 2.220 2 2.130 ₩-2.171 3.645 м 3.646 2 2 0 1 -1 3.128 3.104 0 -2 2.16 ۰t 02 2.154 3.109 S--₩+ 2.083 w. 2 ò 2 2.148 3.080 0 3.079 2.999 2.995 M-ь 3 2.098 -2 0 2 2.094 1 2,988 w. 0 8 2.080 2.904 2.910 2.785 2 2 2.080 2.076 2,785 3 2.773 1 0 0 6 0 2.066 2.767 ₩+ 0 2.020 ν**Ψ**-2.028 3 ~1 ۳. 1.970 2 1 2.745 2.748 2.669 -2 2 -2 1.965 5 1.959 2 2 0 0 0 2.545 2.544 Vπ'--1.920 2.434 2.434 w 2 1.908 1.905 2 2 -1 2.430 3 3 -2 2 2,408 2 2.397 2 1.899 1 0 vw_ 2.397 Ψ 1.877 1.949 2,396 2 -1 [†]The choice between indices bkl and bkl was made in these cases by comparing the intensities with those on the single crystal patterns. The observed spacings have been corrected for absorption error. Specimen radius = 0.15 mm. Table 4. Cs H Cl Maleate

Table 3. Rb H Cl Maleate (Indexing based on body-centered cell)

(Indexing based on body-centered cell)											
Ь	k	1	<i>d</i> _c , Å	d _u , Å	1,	Ь	k	1	d _c , Å	d _o , Å	t _o
0	2	0	8.663	8.676	₩+ •	1	3	2	2.257		
ĩ	ĩ	ō	7.523	7.508	w- •	0	7	1	2.231	2.231	w •
ō	ī	1	4.938	4.932	VW- •	0	4	2	2.214		
1	3	ō	4.750	4.758	W+ •	2	· 0	-2	2.205		
1	ó	-1	4.411	4.414	w •	2	0	2	2,179		
ī	ō	ī	4.359	4.364	w	3	5	0	2.170	2.170	w
ō	4	0	4.331			0	8	0	2.166		
2	0	0	4.176	4.183	₩- •	3	4	-1	2.141	2 1 20	777 L
1	2	-1	3.930	3.925	M •	2	2	-2	2.137 🕽	2.139	w- D
1	2	1	3.894	3.894	м •	3	4	1	2.123		
ō	3	1	3.844	3.839	₩- •	2	2	2	2.114	2.113	W- Ь
2	2	0	3.761	3.763	м •	4	0	0	2.088		
2	1	-1	3.209	3.205	м	4	2	0	2.030	2.029	W
1	5	0	3.201			1	5	-2	2.012		
2	1	1	3.169	3.171	M+ *	1.	5.	2	2,002	2.001	₩ b
1	4	1	3.090	3.089	мь	2.	7	-1	1.972		
1	4	1	3.072			2	4	-2	1.965	1 0/4	
2	4	0	3.006	3.004	W+ •	2	7	1	1.963)	1,904	**
0	6	0	2.888	2.888	₩гь	2	4	2	1.947		
ō	ŝ	1	2.875			1 1	8	-1	1.944		
ź	3	~1	2.842	2.842	τ	1	8	1	1.939		
2	3	1	2.814	2.814	₩ •	4	1	-1	1.932		
3	1	0	2.748			2	8	. 0	1.922		
ō	0	2	2.576	2.574	v₩-•	0	6	2	1.922		
3	3	0	2.508	2.505	tr •	4	1	1	1.914		
ō	2	2	2.469	1		3	1	-2	1.892		
3	ō	-1	2.463	2.466	₩Ь	4	4	0	1.881	1 070	
1	1	-2	2.446			1	9	0	1.876	1.8/8	<i>6</i> -
3	ō	1	2.436	2.436	W	3	6	-1	1.874		
ĩ	1	2	2,428			3	1	2	1.867		
ī	6	-1	2.416			3	6	1	1,862		
ī	6	1	2,407	2,403	tr	3	7	0	1.850		
2	š	-1	2.376			4	3	-1	1.842		
2	6	ō	2,375	1		4	3	1	1.827		
1	7	ō	2.373	2.374	мь	3	3	-2	1.808		
3	2	-1	2,369			0	9	1.	1.803	1.803	И.—
2	5	ī	2,360			3	3	2	1.786		
•3	2	- î	2.345	2.345	W.	2	6	-2	1.753		
ĩ	3	-2	2.271	2,274	₩- •	1	7	-2	1.749		

The observed spacings have been corrected for absorption error. Specimen radius = 0.33 mm.

the monoclinic patterns is on the basis of the bodycentered cell.

Observations that were used in the least-squares refinement of the cell.

An investigation of the structure of potassium acid chloromaleate is in progress, but no further work is planned on the rubidium and cesium salts.

The writers wish to express their thanks to Dr B. M. Benjamin of this Laboratory, who prepared the rubidium and cesium salts, to Mr P. A. Agron of this Laboratory, who made the density measurements, and to Miss Alice McSpadden, now a student at Goucher College, who prepared many of the diffraction patterns.

References

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d_o, Å

2.302 Mi

2.280 м

2.230 11 *61

2.151 w--ь

2.081 tr

2.063

2.026

1.970 w__

1.058 w.,.

1.919

1.898 w-

I,

W 2.383

w 2,099

v₩--

•1